Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

David Gravestock* and Jean M. McKenzie

School of Chemical and Physical Sciences, University of Natal, Pietermaritzburg, Private Bag X01, Scottsville 3209, South Africa

Correspondence e-mail: gravestockd@nu.ac.za

Key indicators

Single-crystal X-ray study T = 295 K Mean σ (C–C) = 0.005 Å R factor = 0.085 wR factor = 0.287 Data-to-parameter ratio = 15.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(5*S**,6*R**,7*R**,9*R**)-1-Benzyl-7,9-dimethyl-8-oxo-1-azaspiro[4.4]nonane-6-carbonitrile

The racemic title compound, $C_{18}H_{22}N_2O$, isolated from the reaction of 2-(1-benzyl-2-pyrrolidinylidene)acetonitrile, 2,4-dibromo-3-pentanone and nonacarbonyldiiron, crystallizes in a centrosymmetric monoclinic space group with four symmetry-equivalent molecules per unit cell. All bond lengths and angles are within expected ranges.

Received 18 November 2002 Accepted 27 November 2002 Online 19 December 2002

Comment

The 1-azaspiro[4.4]nonane ring system is embedded in alkaloids such as cephalotaxine (Fig. 1). Certain esters of cephalotaxine, first discovered by Paudler *et al.* (1963), display antitumour activity and thus are important synthetic targets. Noyori and co-workers showed that polybromoketones undergo [3 + 2]-cycloaddition reactions with 1,3-dienes (Takaya *et al.*, 1978), aromatic olefins (Hayakawa *et al.*, 1978*a*) and enamines (Hayakawa *et al.*, 1978*b*) in the presence of Fe₂(CO)₉. We wished to investigate whether the Noyori annulation reaction could be used to couple 2-methylenepyrrolidines to α, α' -dibromoketones, to form 8-oxo-1azaspiro[4.4]nonanes, synthetic precursors to the cephalotaxine azaspirocycle.



Preparation of the title compound involved heating a mixture of 2-(1-benzyl-2-pyrrolidinylidene)acetonitrile, (1), 2,4-dibromo-3-pentanone, (2), and diiron nonacarbonyl for 18 h with concomitant irradiation, as shown in the reaction scheme. The title compound, (3), selectively crystallized from an EtOAc/hexane solution to give large colourless crystals. We have characterized this compound using HRMS and NMR and FT–IR spectroscopy. The crystal structure of (3) is reported here.



© 2003 International Union of Crystallography Printed in Great Britain – all rights reserved



Figure 2

The molecular structure of (3), showing the labelling scheme. Displacement ellipsoids are drawn at the 50% probability level for all non-H atoms; spheres of arbitrary radii depict H atoms.



Figure 3 Packing diagram, viewed down the *a* axis.

The X-ray crystal structure confirms that the substance produced is a racemic mixture of (5S,6R,7R,9R)- and (5R,6S,7S,9S)-1-benzyl-7,9-dimethyl-8-oxo-1-azaspiro[4.4]-nonane-6-carbonitrile (Fig. 2), as predicted by NOESY NMR experiments. Fig. 3 shows a packing diagram. There are no significant intermolecular interactions.

Experimental

2-(1-Benzyl-2-pyrrolidinylidene)acetonitrile, (1) (0.85 g, 4.3 mmol), dissolved in benzene (10 ml), and 2,4-dibromo-3-pentanone, (2) (1.57 g, 6.44 mmol, passed through a basic alumina column before use), dissolved in benzene (10 ml), were added to $Fe_2(CO)_9$ (1.49 g, 4.10 mmol). The resulting mixture was stirred overnight under N₂ at 323 K, with irradiation (a 400 W high-pressure Hg lamp was used, with an aqueous CuSO₄ solution (10% w/v) functioning as a filter, thereby blocking wavelengths less than 350 nm). The solution was diluted with EtOAc (30 ml) and then washed with saturated aqueous

NaHCO₃ (40 ml) followed by a brine solution (40 ml). The organic layer was separated and dried (MgSO₄), filtered and concentrated in vacuo. A complete separation of all the products proved impossible even after extensive purification by radial chromatography (10% EtOAc/hexane). The title compound selectively crystallized from an EtOAc/hexane solution as large colourless crystals; $R_F 0.49$ (EtOAchexane, 20%); m.p. 381.5-383.5 K. Spectroscopic analysis, IR (KBr, ν_{max} /cm⁻¹): 3100–2800 (CH), 2237 (C=N), 1747 (C=O) and 1076 (C-N); ¹H NMR (500 MHz; CDCl₃, p.p.m.): 7.25-7.35 (5H, m, aromatic H), 3.84 (1H, d, J = 13.3 Hz, ArCH_aH_b), 3.81 (1H, d, J =13.3 Hz, ArCH_aH_b), 2.98-3.03 (1H, m, NCH_aH_bCH₂), 2.80-2.84 (1H, *m*, NCH_a*H*_bCH₂), 2.75 (1H, *d*, *J* = 11.4 Hz, CHC=N), 2.36–2.46 (2H, $m, 2 \times CH_3CH$), 1.99–2.11 (2H, $m, CH_2CH_aH_bCH_2$ and NCCH_aH_b), 1.88-1.93 (1H, m, NCCH_aH_b), 1.75-1.85 (1H, m, CH₂CH_aH_bCH₂), 1.27 (3H, d, J = 7.3 Hz, CHCHCH₃) and 1.14 (3H, d, J = 6.9 Hz, CCHCH₃); ¹³C NMR (125 MHz; CDCl₃, p.p.m.): 213.71 (C=O), 138.94 (Ar C-1'), 128.52, 127.99, 127.29 (Ar C-2', C-3', C-4'), 119.69 $(C \equiv N)$, 71.08 (C - N), 52.17 (NCH_2CH_2) , 51.72 (CH_2Ar) , 49.44 (CCHCH₃), 46.40 (CHCHCH₃), 40.88 (CHC=N), 29.76 (NCCH₂), 22.16 (CH₂CH₂CH₂), 14.06 (CHCHCH₃) and 8.70 (CCHCH₃); m/z (EI): 282 (18%, M⁺), 200 (13), 188 (14), 187 (100), 186 (28) and 91 (71, CH₂Ar); found: *M*⁺ 282.1734; C₁₈H₂₂N₂O requires 282.1732.

Crystal data

 $C_{18}H_{22}N_2O$ $M_r = 282.38$ Monoclinic, $P2_1/n$ a = 10.543 (3) Å b = 10.614 (4) Å c = 14.570 (3) Å $\beta = 93.68$ (2)° V = 1627.2 (8) Å³ Z = 4

Data collection

Enraf-Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: none 6534 measured reflections 2863 independent reflections 2336 reflections with $I > 2\sigma(I)$ $R_{int} = 0.067$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.085$ $wR(F^2) = 0.287$ S = 1.042863 reflections 190 parameters H-atom parameters constrained $D_x = 1.153 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation Cell parameters from 25 reflections $\theta = 2-12^{\circ}$ $\mu = 0.07 \text{ mm}^{-1}$ T = 295 (2) K Rectangular block, colourless $0.8 \times 0.6 \times 0.5 \text{ mm}$

```
\begin{array}{l} \theta_{\max} = 25.0^{\circ} \\ h = -1 \rightarrow 12 \\ k = -12 \rightarrow 12 \\ l = -17 \rightarrow 17 \\ 3 \text{ standard reflections} \\ \text{frequency: } 120 \text{ min} \\ \text{intensity decay: } 3\% \end{array}
```

$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.1715P)^2 \\ &+ 0.8378P] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} &= 0.013 \\ \Delta\rho_{\text{max}} &= 0.90 \text{ e } \text{ Å}^{-3} \\ \Delta\rho_{\text{min}} &= -0.38 \text{ e } \text{ Å}^{-3} \end{split}$$

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3* for Windows (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

We thank the South African National Research Foundation (NRF) and the University of Natal for providing the funding for the research. JMM also thanks the NRF for a grant-holder bursary. We thank Dr O. Q. Munro and Mr J. Ryan, both of the University of Natal, for data collection and solving the crystal

structure, and Mr M. Watson (University of Natal) and Dr P. R. Boshoff (Cape Technikon) for recording NMR spectra and mass spectra, respectively.

References

Enraf-Nonius (1994). *CAD-4 EXPRESS Software*. Enraf-Nonius, Delft, The Netherlands.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany. Hayakawa, Y., Yokoyama, K. & Noyori, R. (1978a). J. Am. Chem. Soc. 100, 1791–1799.
- Hayakawa, Y., Yokoyama, K. & Noyori, R. (1978b). J. Am. Chem. Soc. 100, 1799–1806.
- Paudler, W. W., Kerley, G. I. & McKay, J. (1963). J. Org. Chem. 28, 2194–2197. Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Takaya, H., Makino, S., Hayakawa, Y. & Noyori, R. (1978). J. Am. Chem. Soc. 100, 1765–1777.